9.5 Review Questions

1. What are the shapes of the following molecules?						
	BeCl ₂	BF ₃	CBr ₄	K ₂ O	Rbl	
2. Make the	e electron dot	for C ₃ H ₆ O.	What would	d its shape	be?	
3. What shape are the following molecules?						
	a. CCI ₄		d. AIB	d. AlBr ₃		
	b. Na ₂ S		e. H₃N	l		
	c. RbF		f. Li ₂ C)		
4. Both Be and O have two bonding sites. When Be is the central atom, the shape of the molecule is linear. When O is the central atom, the molecule is a bent line. Why are they different?						
5. What sha	ape is a mole	cule of CO ₂	?			