## 9.6-7 Review Questions

1. Tell whether the following molecules are polar or nonpolar.	
a. H <sub>2</sub> O	d. SiBr <sub>4</sub>
b. C <sub>2</sub> H6	e. PH <sub>3</sub>
c. LiBr	f. BH <sub>3</sub>
2. Why can't a molecule made of nonpolar bonds be polar?	
3. What has a higher melting point, a polar molecule or a same-sized nonpolar molecule?	
4. Is K <sub>3</sub> P a polar, non-polar, or an ionic molecule? How do you know?	
5. The things listed below will either dissolve in water (polar) or gasoline (nonpolar). Please tell what each will dissolve better in.	
a. A molecule made of Li and	d F d. Sunscreen
b. Motor oil	e. Table salt
c. H <sub>2</sub> O	f. Your hair
6. The Wicked Witch of the West died when Dorothy Gale threw water on her in the movie, The Wizard of OZ. Knowing this, what type of molecules must have held the witch together? WHY?	